

Crystal structures of two polysubstituted derivatives of cyclohexanone

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Received April 24, 2012; accepted May 30, 2012

The structure of 2,4-diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone (I) was determined by X-ray crystallography. The symmetry operation $(-x + 2, -y + 1, -z)$ generates the whole molecule. The compound crystallizes in an orthorhombic system and was characterized thus: P_{bca} , $a = 8.6405(10)$, $b = 18.682(2)$, $c = 20.187(4)$, $\alpha = \beta = \gamma = 90^\circ$, $Z = 8$, $V = 3258.7(8) \text{ \AA}^3$. The crystal structure was solved by direct methods and refined by full-matrix least-squares on F^2 to final values of $R1 = 0.0687$ and $wR2 = 0.1225$. The structure of 2,4-Diacetyl-3-(4-fluorophenyl)-5-hydroxy-5-methylcyclohexanone (II) was determined by X-ray crystallography. The symmetry operation $(-x + 2, -y + 1, -z)$ generates the whole molecule. The compound crystallizes in a monoclinic system and was characterized thus: $P2_1/c$, $a = 5.7045(5)$, $b = 18.120(3)$, $c = 16.1678(11)$, $\alpha = \gamma = 90^\circ$, $\beta = 96.192(6)^\circ$, $Z = 4$, $V = 1661.5(3) \text{ \AA}^3$. The crystal structure was solved by direct methods and refined by full-matrix least-squares on F^2 to final values of $R1 = 0.0699$ and $wR2 = 0.02100$.

Key words: 2,4-Diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone, 2,4-Diacetyl-3-(4-fluorophenyl)-5-hydroxy-5-methylcyclohexanone, 1,3-dicarbonyl compounds, crystal structure, aldol condensation reaction

INTRODUCTION

The investigations on the interaction between 1,3-dicarbonyl compounds (2,4-pentanedione or ethylacetoacetate) and aldehydes or ketones began at the end of the 19th century and are continuing till present [1-4]. The results are diverse. It has been accepted that different condensation products are synthesized at different molar ratios of the reactants, by varying temperature or type of catalyst.

In this research the interaction between some aromatic aldehydes and 2,4-pentanedione at a molar ratio of 1:1, 1:2 or 1:4 in the presence of piperidine as a catalyst and glacial acetic acid was studied. The interaction was performed at room temperature for 2 h (Fig. 1).

EXPERIMENTAL

The crystal structure of the compounds was determined using the single crystal X-ray diffraction method. Data collection was carried out at -60°C using graphite-monochromated MoK_α radiation ($\lambda = 1.5418 \text{ \AA}$) on an ENRAF NONIUS four circle diffractometer. The unit cell was determined and refined using the CAD4-EXPRESS program. A semiempirical absorption correction

was performed using the PLATON/ABS PSI program [5]. The structure was solved with direct methods by SHELXS97 [6] and refined with SHELXL97 [7] by least-squares methods based on F^2 . All non-hydrogen atoms were fully refined and all hydrogen atom positions were taken from the electron density map and refined isotropically. The plots of the molecular structure were made using the DIAMOND program (CRYSTAL IMPACT GbR, Bonn, Germany).

Complete data collection parameters and details of the structure solution and refinement are given in Table 1. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif

RESULTS AND DISCUSSION

The structure of the synthesized compounds was proven by IR, $^1\text{H-NMR}$ and mass-spectral analyses. Single crystals suitable for X-ray structure analysis could be obtained by crystallization from an ethanol solution, which afforded colorless crystals. The crystal and structure-refinement data are summarized in Tables 1 – 7.

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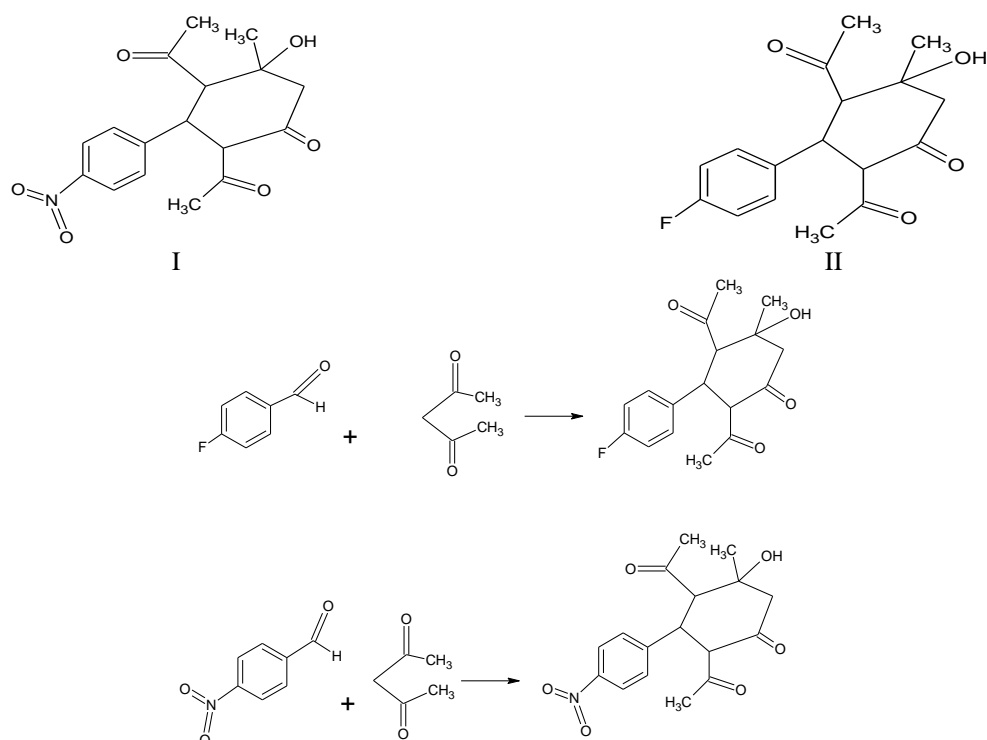


Fig.1. Chemical formula of the title compounds

Table 1. Crystal and experimental data for 2,4-Diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone (I) and 2,4-Diacetyl-3-(4-fluorophenyl)-5-hydroxy-5-methylcyclohexanone (II).

	I	II
Chemical formula	C ₁₇ H ₁₉ NO ₆	C ₁₇ H ₁₉ FO ₄
Formula weight	333.33	306.32
Temperature	213 (2) K	223 (2) K
Crystal system:	Orthorhombic	Monoclinic
Space group:	Pbca	P 2 ₁ /n
	a = 8.6405(10) Å	a = 5.7045(5) Å
	b = 18.682 (2) Å	b = 18.120 (3) Å
	c = 20.187 (4) Å	c = 20.1
	α = β = γ = 90°	β = 96.192 (6)°
		α = γ = 90°
Volume	3258.7 (8) Å ³	1661.5 (3) Å ³
Z	8	4
Density (calculated)	1.359 Mg/m ³	1.225 Mg/m ³
Radiation :	MoKα (λ = 0.71073 Å)	MoKα (λ = 1.5418 Å)
F(000)	1408	648
Crystal size	0.30 x 0.15 x 0.15 mm	.50 x 0.15 x 0.15 mm
No. of reflections collected	2205	3998
No. of independent reflections	1673	2827
θ range for data collection [°]	5.83 to 20.81	5.50 to 64.96
Data / restraints / parameters	1673 / 0 / 273	2827 / 0 / 276
Goodness-of-fit on F ²	1.078	1.067
R indices [I > 2σ (I)]	R1 = 0.0687, wR2 = 0.1225	R1 = 0.0699, wR2 = 0.2100
R indices (all data)	R1 = 0.2550, wR2 = 0.1590	R1 = 0.0792, wR2 = 0.2210
Refinement:	Full matrix	Full matrix
	(Δ/σ) _{max} = 0.001	
	(Δ/ρ) _{max} = 0.38 eÅ ⁻³	
	(Δ/ρ) _{min} = -0.21 eÅ ⁻³	
Measurement:	Enraf-Nonius CAD4	Enraf-Nonius CAD4
Program system:	Enraf-Nonius SDP	Enraf-Nonius SDP
Structure determination:	MULTAN	MULTAN
CCDC deposition number:	706401	706402

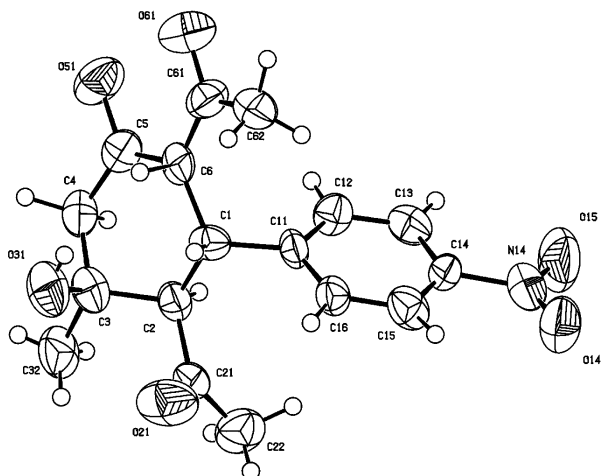


Fig. 2. ORTEP structure of 2,4-diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone (I), showing 50% probability ellipsoids.

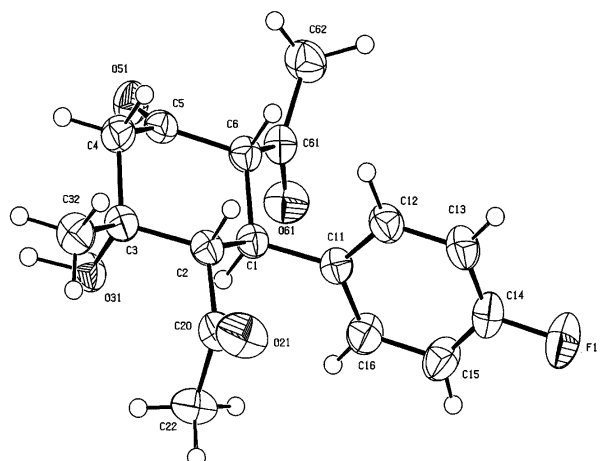


Fig. 3. ORTEP structure of 2,4-diacetyl-3-(4-fluorophenyl)-5-hydroxy-5-methylcyclohexanone (II), showing 50% probability ellipsoids.

The interaction of 4-bromobenzaldehyde, 4-hydroxybenzaldehyde and 4-methoxybenzaldehyde with 2,4-pentandione results in the formation of the respective 3-(4-substituted-benzylidene)pentane-2,4-dione [8]. We established that the condensation process between aromatic aldehydes (containing a nitro group or a fluorine atom at para-position of the aromatic nucleus compared to aldehyde group) and 1,4-pentandione produced polysubstituted derivative of cyclohexanone (fig.: 2 and 3). These two substituents reveal -I-effect and strong -M-effect. X-ray crystal analysis proved a carbocyclic product, a result of the interaction between one mole aldehyde and two moles 2,4-pentandione, namely 2E-4-diacetyl-5-hydroxy-5-methyl-3-(4'-nitro-phenyl)-cyclohexanone or 2E-4-diacetyl-5-hydroxy-5-methyl-3-(4'-fluorophenyl)-cyclohexanone

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for 2,4-diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone (I).

Atom	x	y	z	U (eq)
C 1	2663 (10)	1890 (5)	3991 (4)	47 (2)
C 2	3546 (10)	2560 (4)	3769 (5)	47 (2)
C 3	2533 (12)	3219 (5)	3687 (4)	57 (2)
C 4	1370 (15)	3065 (6)	3149 (5)	62 (3)
C 5	580 (12)	2362 (5)	3231 (4)	54 (2)
C 6	1106 (10)	1837 (5)	3653 (4)	47 (2)
C 11	3689 (9)	1251 (4)	3912 (4)	43 (2)
C 12	3860 (12)	927 (5)	3308 (4)	51 (3)
C 13	4914 (12)	378 (5)	3198 (5)	56 (3)
C 14	5784 (9)	176 (4)	3732 (6)	52 (2)
C 15	5664 (12)	488 (5)	4338 (5)	59 (3)
C 16	4609 (11)	1021 (5)	4425 (4)	48 (2)
C 21	4871 (12)	2699 (4)	4242 (5)	52 (2)
C 22	6485 (14)	2644 (7)	3973 (7)	75 (3)
C 32	3483 (17)	3863 (6)	3518 (7)	77 (4)
C 61	123 (12)	1223 (5)	3760 (5)	59 (2)
C 62	421 (10)	709 (4)	4314 (4)	70 (3)
O 14	7672 (8)	-617 (3)	4112 (4)	93 (2)
O 15	7166 (9)	-637 (3)	3075 (4)	112 (3)
O 21	4639 (7)	2839 (3)	4825 (3)	94 (2)
O 31	1678 (7)	3386 (3)	4293 (3)	79 (2)
O 51	-722 (8)	2282 (3)	2893 (3)	80 (2)
O 61	-1099 (7)	1135 (3)	3427 (3)	84 (2)
N 14	6962 (11)	-393 (4)	3626 (5)	77 (3)

Table 3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for 2,4-diacetyl-3-(4-fluorophenyl)-5-hydroxy-5-methylcyclohexanone (II)

Atom	x	y	z	U (eq)
C 1	455 (4)	7077 (1)	11011 (2)	33 (1)
C 2	2355 (4)	7048 (1)	10391 (2)	33 (1)
C 3	2335 (4)	6307 (1)	9931 (2)	35 (1)
C 4	2871 (5)	5686 (2)	10577 (2)	38 (1)
C 5	1031 (4)	5701 (1)	11171 (2)	35 (1)
C 6	816 (4)	6427 (1)	11629 (2)	34 (1)
C 11	498 (5)	7812 (1)	11459 (2)	34 (1)
C 12	2456 (5)	8043 (2)	11982 (2)	42 (1)
C 13	2494 (6)	8726 (2)	12373 (2)	47 (1)
C 14	550 (5)	9167 (2)	12231 (2)	46 (1)
C 15	-1407 (6)	8966 (2)	11727 (2)	50 (1)
C 16	-1433 (5)	8281 (2)	11339 (2)	44 (1)
C 20	2182 (5)	7707 (2)	9803 (2)	40 (1)
C 22	-66 (6)	7860 (2)	9261 (2)	54 (1)
C 32	4110 (5)	6290 (2)	9288 (2)	45 (1)
C 61	-1143 (5)	6363 (1)	12196 (2)	38 (1)
C 62	-541 (7)	5979 (2)	13006 (2)	52 (1)
O 21	3891 (4)	8097 (1)	9779 (2)	63 (1)
O 31	-3 (3)	6195 (1)	9531 (1)	39 (1)
O 51	-264 (4)	5184 (1)	11272 (1)	45 (1)
O 61	-3095 (4)	6606 (1)	11985 (1)	51 (1)
F 1	582 (4)	9845 (1)	12605 (1)	66 (1)

Table 4. Selected bond lengths (Å) for 2,4-diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone (I)

Bond	Length	Bond	Length
C 1 – C 11	1.496 (10)	C 5 – C 6	1.377 (10)
C 1 – C 6	1.512 (10)	C 6 – C 61	1.444 (11)
C 1 – C 2	1.532 (10)	C 14 – N 14	1.487 (11)
C 2 – C 21	1.513 (10)	C 21 – O 21	1.223 (9)
C 2 – C 3	1.521 (10)	C 21 – C 22	1.500 (12)
C 3 – O 31	1.462 (8)	C 61 – O 61	1.261 (9)
C 3 – C 32	1.495 (12)	C 61 – C 62	1.496 (10)
C 3 – C 4	1.506 (11)	O 14 – N 14	1.231 (8)
C 4 – C 5	1.490 (12)	O 15 – N 14	1.214 (8)
C 5 – O 51	1.323 (9)		

Table 5. Selected bond angles (°) for 2,4-diacetyl-5-hydroxy-5-methyl-3-(4-nitrophenyl)-cyclohexanone (I)

Bond	Angle	Bond	Angle
C 11 – C 1 – C 6	115.2 (7)	C 5 – C 6 – C 61	117.7 (9)
C 11 – C 1 – C 2	109.0 (7)	C 5 – C 6 – C 1	121.8 (9)
C 6 – C 1 – C 2	111.4 (8)	C 61 – C 6 – C 1	120.5 (9)
C 21 – C 2 – C 3	111.4 (7)	C 16 – C 11 – C 1	120.7 (8)
C 21 – C 2 – C 1	109.4 (7)	C 12 – C 11 – C 1	120.8 (8)
C 3 – C 2 – C 1	114.0 (8)	C 15 – C 14 – N 14	119.3 (10)
O 31 – C 3 – C 32	107.3 (8)	C 13 – C 14 – N 14	117.4 (10)
O 31 – C 3 – C 4	107.9 (8)	O 21 – C 21 – C 22	121.0 (9)
C 32 – C 3 – C 4	110.8 (9)	O 21 – C 21 – C 2	121.3 (9)
O 31 – C 3 – C 2	111.8 (7)	C 22 – C 21 – C 2	117.6 (9)
C 32 – C 3 – C 2	111.1 (10)	O 61 – C 61 – C 6	121.0 (9)
C 4 – C 3 – C 2	107.9 (8)	O 61 – C 61 – C 62	117.3 (9)
C 5 – C 4 – C 3	113.3 (9)	C 6 – C 61 – C 62	121.4 (9)
O 51 – C 5 – C 6	121.3 (9)	O 15 – N 14 – O 14	122.0 (9)
O 51 – C 5 – C 4	115.6 (9)	O 15 – N 14 – C 14	120.0 (10)
C 6 – C 5 – C 4	123.0 (10)	O 14 – N 14 – C 14	118.0 (9)

Table 6. Selected bond lengths (Å) for 2,4-diacetyl-3-(4-fluorophenyl)-5-hydroxy-5-methylcyclohexanone (II)

Bond	Length	Bond	Length
C 1 – C 11	1.515 (3)	C 5 – O 51	1.214 (3)
C 1 – C 6	1.543 (3)	C 5 – C 6	1.520 (3)
C 1 – C 2	1.555 (3)	C 6 – C 61	1.525 (4)
C 2 – C 20	1.522 (4)	C 14 – F 1	1.367 (3)
C 2 – C 3	1.535 (3)	C 20 – O 21	1.208 (4)
C 3 – O 31	1.432 (3)	C 20 – C 22	1.499 (4)
C 3 – C 32	1.528 (4)	C 61 – O 61	1.211 (3)
C 3 – C 4	1.544 (4)	C 61 – C 62	1.490 (4)
C 4 – C 5	1.498 (4)		

Table 7. Selected bond angles (°) for 2,4-diacetyl-5-hydroxy-5-methyl-3-(4-fluorophenyl)-cyclohexanone (II)

Bond	Angle	Bond	Angle
C 11 – C 1 – C 6	111.61 (19)	C 4 – C 5 – C 6	115.2 (2)
C 11 – C 1 – C 2	111.2 (2)	C 5 – C 6 – C 61	109.2 (2)
C 6 – C 1 – C 2	109.7 (2)	C 5 – C 6 – C 1	111.0 (2)
C 20 – C 2 – C 3	112.8 (2)	C 61 – C 6 – C 1	113.1 (2)
C 20 – C 2 – C 1	111.9 (2)	C 16 – C 11 – C 1	120.0 (2)
C 3 – C 2 – C 1	111.9 (2)	C 12 – C 11 – C 1	121.7 (2)
O 31 – C 3 – C 32	110.1 (2)	C 15 – C 14 – F 1	118.5 (3)
O 31 – C 3 – C 2	107.49 (19)	C 13 – C 14 – F 1	118.7 (3)
C 32 – C 3 – C 2	112.1 (2)	O 21 – C 20 – C 22	121.1 (3)
O 31 – C 3 – C 4	108.0 (2)	O 21 – C 20 – C 2	118.7 (3)
C 32 – C 3 – C 4	110.4 (2)	C 22 – C 20 – C 2	120.2 (2)
C 2 – C 3 – C 4	108.7 (2)	O 61 – C 61 – C 62	122.3 (3)
C 5 – C 4 – C 3	108.6 (2)	O 61 – C 61 – C 6	121.0 (2)
O 51 – C 5 – C 4	123.4 (2)	C 62 – C 61 – C 6	116.7 (2)
O 51 – C 5 – C 6	121.4 (2)		

Florencio et al.[9] established that during the reaction between nickel acetylacetonate dihydrate and p-nitrobenzyl chloride with dimethyl sulfoxide as a solvent only the monoalkylation product was obtained instead of the expected mixture of dialkylation and monoalkylation products. In this reaction a compound with empirical formula $C_{17}H_{19}NO_6$ was obtained, which was analyzed by X-ray diffraction methods in order to obtain conformational data. The product was 2E-4-diacetyl-5Z-methyl-3-(p-nitrophenyl)-cyclohex-1-en-1,5E-diol. The unit cell dimensions reported by Florencio were: $a=20.176(2)$, $b=8.623(3)$, $c=18.746(9)$ Å [9]. Our data do not correspond to their data: $a=8.6405(10)$, $b=18.682(2)$ and $c=20.187(4)$ Å. analogous results were obtained using 4-fluorobenzaldehyde. The bond lengths were longer than the respective ones of the compound containing a nitro group. It seems that the electronegativity and the mesomeric effect of the nitro group is due to the difference in the bond length. The two cyclohexan rings were in a distorted half-chair conformation. The acetyl group was attached to C2 in a trans-equatorial position and the hydroxyl group attached to C3 was situated in an axial position for both compounds. Ponomarev et al.[10] described the synthesis of a cyclic product with the same molecular formula based only on the data of a spectral analysis as well as data of elemental analysis for carbon, hydrogen and

fluorine. However, the results of the elemental analysis do not correspond to the formula suggested by them. The synthesized by us substance corresponds to the suggested formula. The results are based on the spectral, elemental and X-ray crystal structure analysis.

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КРИСТАЛНИ СТРУКТУРИ НА ДВЕ ПОЛИСУБСТИТУИРАНИ ПРОИЗВОДНИ НА ЦИКЛОХЕКСАНОНА

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Постъпила на 21 април, 2012 г.; коригирана на 19 септември, 2012 г.

(Резюме)

Структурата на 2,4-диацетил-5-метил-5-хидрокси-3-(4-нитрофенил)циклохексанон (I) е определена чрез монокристален рентгеноструктурен анализ. Съединението кристализира в орторомбична кристална система и пространствена група P_{cn} , с параметри на елементарната клетка $a=8.6405(10)$, $b=18.682(2)$, $c=20.187(4)$, $\alpha=\beta=\gamma=90^\circ$, $Z=8$, $V=3258.7(8)$ Å³. Кристалната структура е доказана с директни методи и точно определена с помощта на метода на най-малките квадрати за F^2 до стойност $R1=0.0687$. Структурата на 2,4-диацетил-3-(4-флуорофенил)-5-хидрокси-5-метилциклохексанон (II) е определена чрез монокристален рентгеноструктурен анализ. Съединението кристализира в моноклинна кристална система и пространствена група $P2_1/c$, с параметри на елементарната клетка $a=5.7045(5)$, $b=18.120(3)$, $c=16.1678(11)$, $\alpha=\gamma=90^\circ$, $\beta=96.192(6)^\circ$, $Z=4$, $V=1661.5(3)$ Å³. Кристалната структура е доказана с директни методи и точно определена с помощта на метода на най-малките квадрати за F^2 до стойност $R1=0.0699$.